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THE OPEN UNIVERSITY OF SRI LANKA
B.Sc. Degree Programme and Stand Alone Courses in
Science - 2012/2013
CMU2221/CME4221 - Organic Chemistry 1

CONTINUOUS ASSESSMENT TEST 1

Ques. No.	Max.	Marks
1	12	
2	30	
3	24	
4	16	
5	18	
Total	100	

Date: Saturday, 23rd February 2013

Time: 11.00 a.m. – 12.30 p.m.

1. Predict the $\lambda_{\max}$ of the following compound A, using Woodward-Fieser rules for dienes.

Structure A: A bicyclic diene with a methyl group and a carboxylic acid group.	Basic value for Heteroannular diene	= 214 nm	
	Basic value for Homoannular diene	= 253 nm	
	Increments for,		
	Double bond extending conjugation	= +30 nm	
	Alkyl substituent or ring residue	= +05 nm	
	Exocyclic double bond	= +05 nm	
	$\lambda_{\max}$		

(12 Marks)

2. The compound B with the molecular formula $C_9H_{10}O$, gave an orange coloured precipitate with Brady's reagent. It showed a strong IR absorption at $\nu_{\max} 1725 \text{ cm}^{-1}$. Its ^1H NMR spectrum showed following signals.

δ ppm 1.2 triplet (3H) 2.2 quartet (2H) 7.4 two doublets (2H each) 10.2 singlet (1H)

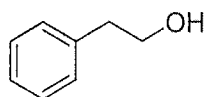
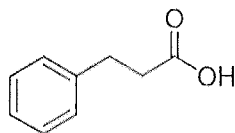
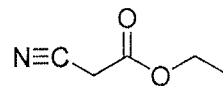
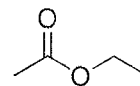
Elucidate the structure of B and assign δ values to the protons in it.

Structure elucidation of B:

Structure and assignment of δ values:

(30 marks)

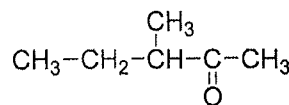
3. IR spectra given in the table belong to **THREE** of the following compounds labelled as **C**, **D**, **E** and **F**. Giving reasons select the correct compound responsible for each spectrum.

**C****D****E****F**

IR spectrum	Compound	Reasons
		
		
		

(24 marks)

4. Draw the fragmentation pattern and the structures of ions responsible for the peaks at m/z 43 and 72 in the EI mass spectrum of compound **G**.

**G**

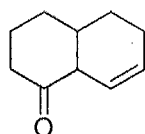
(16 marks)

Reg. No.

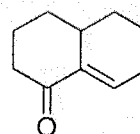
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5. Give **ONE** major difference that you could observe in the **spectra** of each pair of compounds listed below.

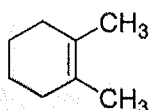
(i) IR spectra of



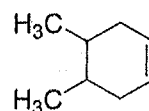
and



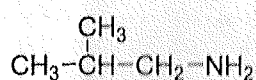
(ii) ^1H NMR spectra of



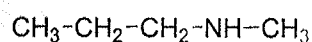
and



(iii) Mass spectra of



and



(18 marks)