

Date: 09th May 2025

Time: 02.00 pm – 04.00 pm

Useful Physical Constants / Conversions

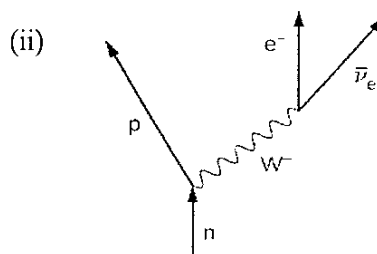
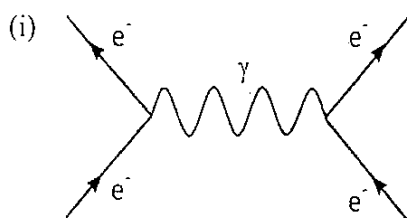
1 amu	= 931.5 MeV/c ²	Mass of $^{14}_7\text{N}$	= 14.003074 amu
Mass of $^1_1\text{H}^+$	= 1.007277 amu	Mass of $^{17}_8\text{O}$	= 16.999132 amu
Mass of $^4_2\text{He}^{++}$	= 4.001506 amu		

Note: Standard symbols have their usual meanings.

Answer Any Four (04) Questions Only

- (01) (a) Briefly explain the ‘liquid drop’ model of the atomic nucleus.
- (b) State **five** key assumptions made in developing the liquid drop model of the nucleus.
- (c) Describe the **successes** and **limitations** of the liquid drop model of the nucleus.
- (d) Briefly describe the Semiempirical Mass Formula.
- (02) (a) Define the terms “half-life” and “mean-life” of a radioactive substance
- (b) Derive an expression for half-life ($T_{1/2}$) of a radioactive element.
- (c) Establish the relationship between the “half-life” and “mean-life”.
- (d) The activity of a radioisotope is found to decrease to one-eighth ($1/8$) of the initial activity in 36 hours. Calculate the following:
- (i) Half-life
- (ii) Disintegration constant
- (03) (a) State two major differences between a chemical reaction and a nuclear reaction.
- (b) The nuclear reaction between an alpha particle and a stationary nitrogen ($^{14}_7\text{N}$) atom results in the transmutation of nitrogen into an oxygen isotope ($^{17}_8\text{O}$), with the emission of a proton.
- (i) Write the equation for the above nuclear reaction (use standard symbols).
- (ii) Compute the **minimum** kinetic energy of the alpha particle for the above reaction to happen (state your answer in MeV).
- (iii) State assumptions (if any) made in computing the above **minimum** kinetic energy.

- (04) (a) Name three (03) types of radiation detection devices.
- (b) Briefly describe the working principle of **anyone** (01) of the three radiation detection devices listed by you above in part (a).
- (c) Discuss the strengths and weaknesses of the device described by you above in part (b).
- (05) (a) Briefly describe the conservation laws for elementary particle reactions.
- (b) Feynman diagrams are graphical representations that show how elementary particles interact. Briefly describe **important rules** followed in constructing a Feynman diagram.
- (c) Describe the elementary particle interactions represented by the following two Feynman diagrams.



- (06) (a) Explain the principle and working of a *Betatron*, with the help of suitable sketches.
- (b) Derive the following *Betatron condition* for a stable orbit of constant radius r in a magnetic induction B caused by the magnetic flux ϕ .
- $$\phi = 2\pi r^2 B$$
- (c) Discuss the different functions of the time varying magnetic field in the *Betatron*.
