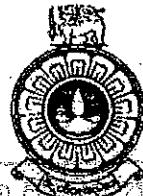


0093

THE OPEN UNIVERSITY OF SRI LANKA  
FACULTY OF HEALTH SCIENCES  
DEPARTMENT OF PHARMACY  
ACADEMIC YEAR 2019/2020 – SEMESTER I



THE OPEN UNIVERSITY OF  
SRI LANKA

22 SEP 2020

EXAMINATION DIVISION

BACHELOR OF PHARMACY HONOURS  
FMU5403- PHARMACEUTICAL TECHNOLOGY I - LEVEL 5  
FINAL EXAMINATION  
DURATION: THREE HOURS

DATE: 22<sup>ND</sup> SEPTEMBER 2020

TIME: 1.30 PM- 4.30 PM

INDEX NO: .....

**Index No.....**

**Part – B (20 marks)**

1.

1.1 Briefly explain three (03) different types of measurements of humidity. (03 marks)

I. ....

.....

II. ....

.....

III. ....

.....

1.2 List two (02) methods of controlling drying rate of drum drier. (02 marks)

I. ....

II. ....

1.3 Briefly explain the term ‘convective drying’ (03 marks)

.....

.....

.....

.....

.....

.....

.....

.....

IV. ....

.....

**Index No.....**

2.

2.1. State three (03) types of mixtures based on mixing behavior. (03 marks)

- I. ....
- II. ....
- III. ....

2.2. Briefly describe the effect of particle size in powder segregation. (03 marks)

.....  
.....  
.....  
.....  
.....  
.....  
.....

2.3. State four (04) methods to reduce powder segregation. (02 marks)

- I. ....
- II. ....
- III. ....
- IV. ....

2.4. State four (04) mechanisms involved in liquid-liquid mixing. (02 marks)

- I. ....
- II. ....
- III. ....
- IV. ....

**Index No.....**

**Part C – (60 marks)**

**Answer all questions.**

1.

- 1.1. List two (02) mechanisms of heat transfer. (01 mark)
- 1.2. Write a short note on the following. (06 marks)
  - 1.2.1. Heat exchangers
  - 1.2.2. Insulating materials
- 1.3. Calculate the rate of heat transfer through a flat copper plate of 100 mm tall, 150 mm wide and 25 mm thick. The surface temperatures are 100 °C and 50 °C. Thermal Conductivity (k) of copper is 385 W/mK. (03 marks)
- 1.4.
  - 1.4.1. State two (02) types of mechanisms in filtration. (01 mark)
  - 1.4.2. Briefly explain four (04) methods used to increase filtration rate. (04 marks)

2.

- 2.1. According to Reynold's equation, state four (04) factors that influence fluid flow. (02 marks)
- 2.2. Briefly explain three (03) types of fluid flow explained by Reynolds experiment using diagrams. (06 marks)
- 2.3.
  - 2.3.1. Define the term crystallization. (01 mark)
  - 2.3.2. Briefly explain differences of crystalline and amorphous solids. (04 marks)
- 2.4. State two methods to achieve supersaturation. (02 marks)

**Index No.....**

3.

- 3.1. State four (04) factors affecting the rate of evaporation. (02marks)
- 3.2. Answer the following questions with regards to hammer mill.
- 3.2.1. Name four (04) parts of the machine. (02 marks)
- 3.2.2. Explain the principle/s of size reduction process of the above machine. (02 marks)
- 3.2.3. List two (02) factors that determine the size and shape of the milled particles obtained by the above machine. (02 marks)
- 3.3.
- 3.3.1. State two (02) methods used for particle separation. (01 mark)
- 3.3.2. Briefly explain agitation method used for particle separation. (02 marks)
- 3.4. Explain the mechanism of settling chamber machine used for size separation. (04 marks)

4.

- 4.1. List four (04) colligative properties of a solution. (02 marks)
- 4.2. State the Raoult's law. (02 marks)
- 4.3. A liquid mixture consists of two (02) components, namely, A and B. Mole fraction of A in this liquid mixture is 0.8. The mixture is distilled at temperature of 60 °C. The vapour pressures exerted by A and B in their pure state at 60 °C is 10 atm and 30 atm respectively. Assuming this mixture follows Raoult's law, answer the following.
- 4.3.1. Calculate partial pressure ( $P_A$ ) exerted by A. (01 mark)
- 4.3.2. Calculate partial pressure ( $P_B$ ) exerted by B. (01 mark)
- 4.3.3. Calculate mole fractions of A and B in the vapor. (01 mark)
- 4.3.4. Which compound has the lower boiling point? Explain your answer. (02 marks)
- 4.4. Briefly describe the following,
- 4.4.1. Efflorescence
- 4.4.2. Deliquescence
- 4.4.3. Exsiccation (06 marks)

